



NEEST

NEW ENERGY & ENVIRONMENTAL  
SOLUTIONS AND TECHNOLOGIES

# TETHYS WEBINAR - GREEN HYDROGEN PRODUCTION

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# HYDROGEN PRODUCTION



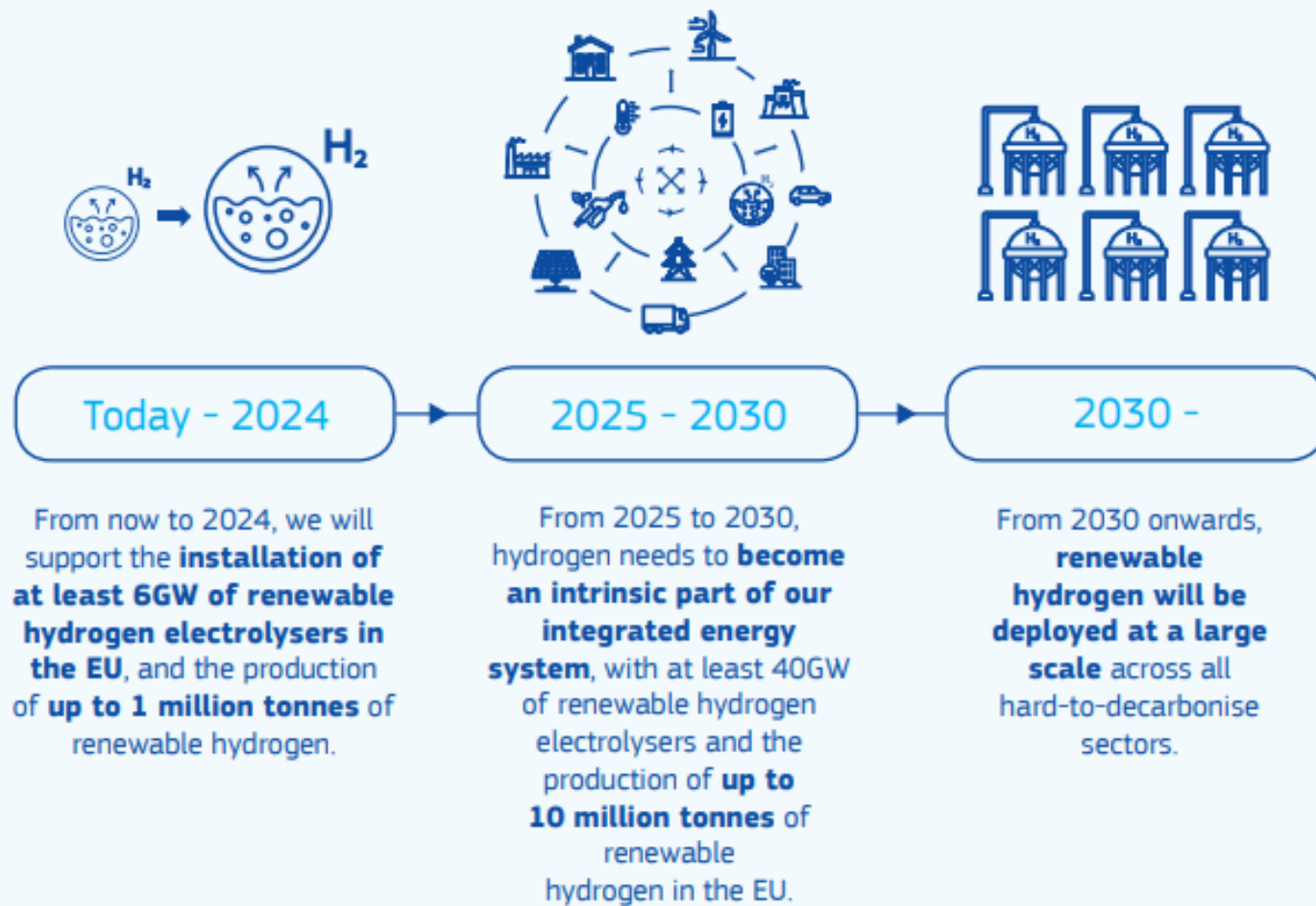
In 1874, in his famous novel *The Mysterious Island*, Jules Verne prophesied the advent of hydrogen in response to the exhaustion of coal: *"I believe that water will one day be used as a fuel, that hydrogen and oxygen, which constitute it, used alone or simultaneously, will provide an inexhaustible source of heat and light of an intensity that coal cannot have."*

- Cyrus Harding, 1874  
hero of Jules Verne's novel  
"*The Mysterious Island*"

150 years later, at a time when the climate emergency has never been so urgent, this energy vector is at the heart of all attention and is enjoying unprecedented acceleration.

# EU Position on Hydrogen

The path towards a European hydrogen eco-system step by step :



How can hydrogen be promoted in Europe?



- The production of clean hydrogen needs to be increased **by creating a sustainable industrial value chain**.



- We should **boost the demand for clean hydrogen** coming from industrial applications and mobility technologies.



- Clean hydrogen needs a **supportive framework, well-functioning markets** and **clear rules**, as well as dedicated infrastructure and a logistical network.



- **Promoting research** and **innovation** in clean hydrogen technologies is crucial.



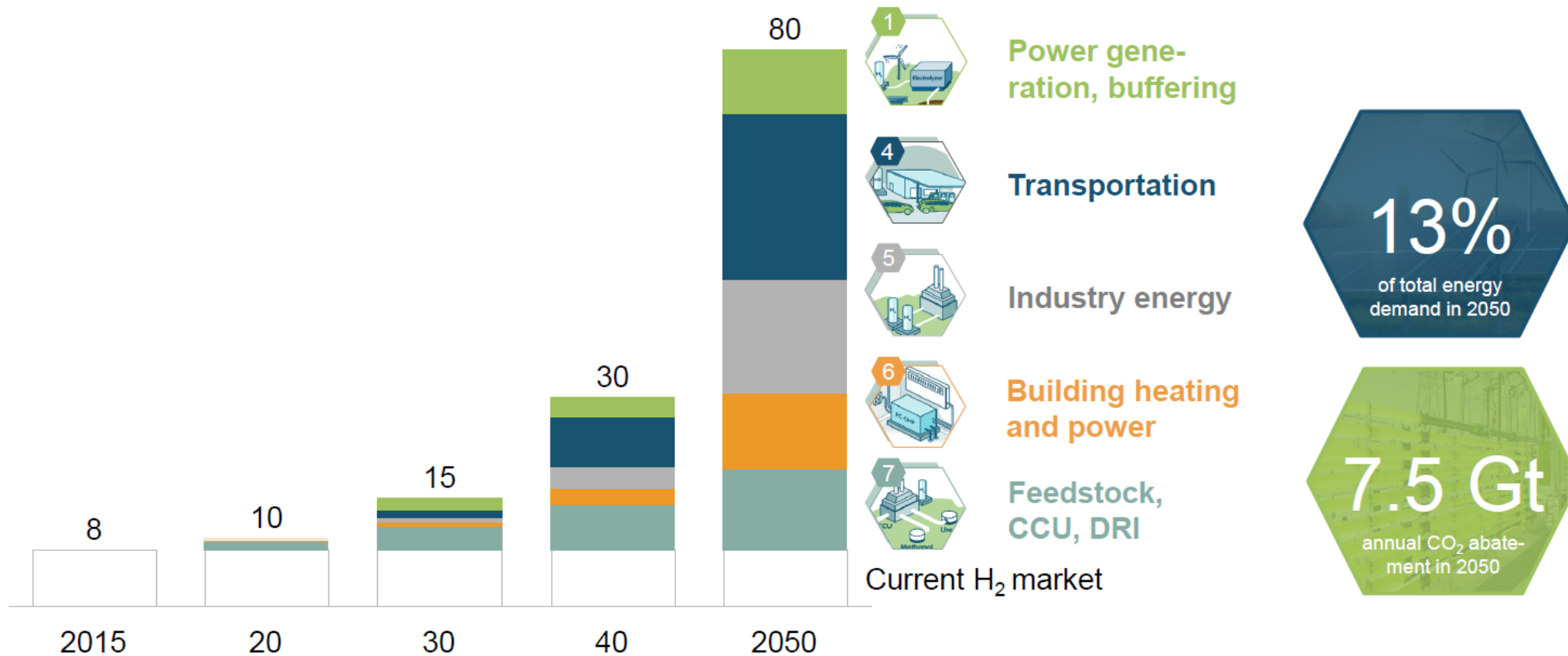
- Europe we will secure **cooperation opportunities with neighboring countries and regions of the EU** and work to establish a global hydrogen market.



- The **European Clean Hydrogen Alliance** will help build up a robust pipeline of investments.

# By 2050, hydrogen can enable major CO<sub>2</sub> emission reductions

Global Energy demand supplied with hydrogen, Exajoule (EJ)



1 Excluding feedstock

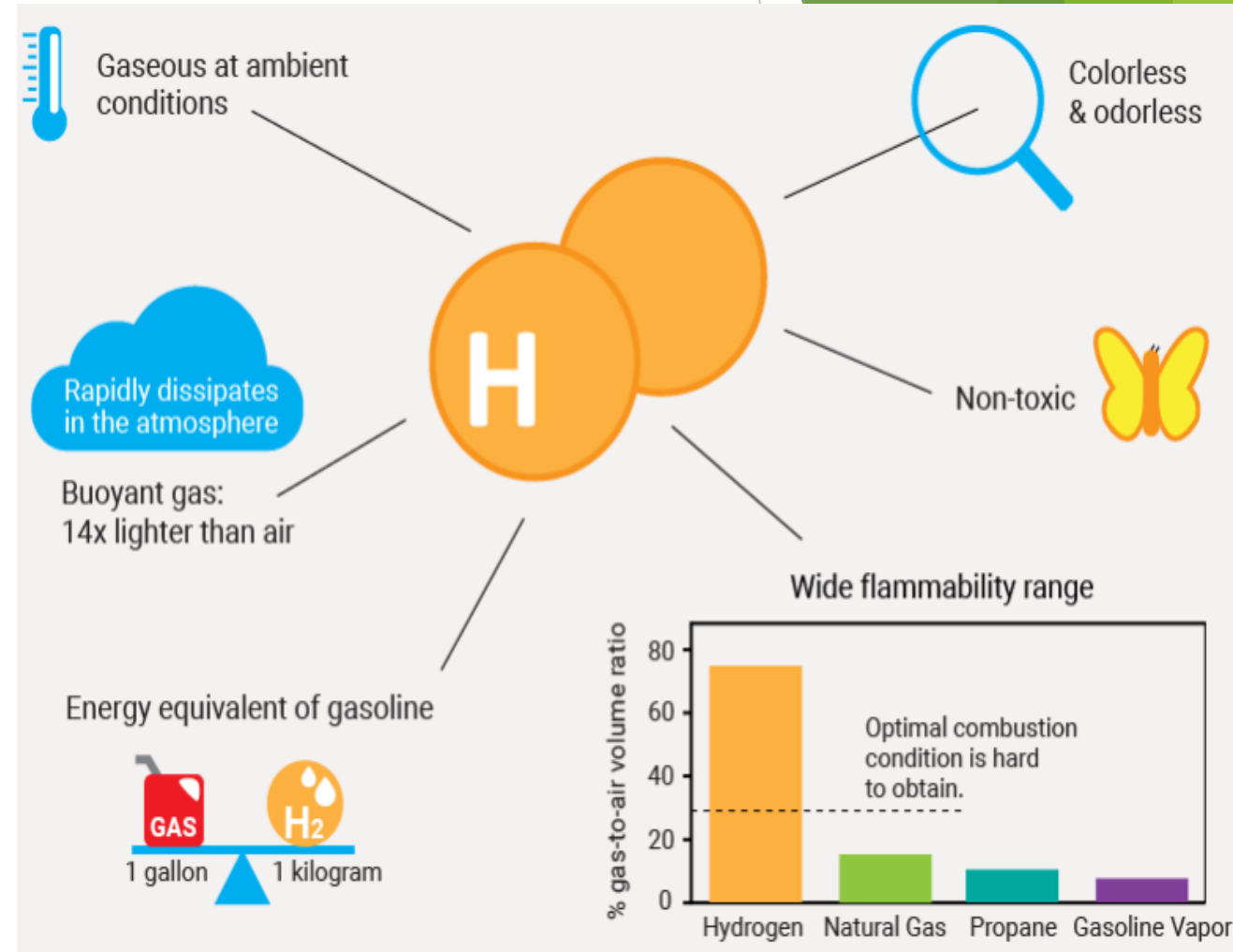
SOURCE: Hydrogen Council, IEA ETP Hydrogen and Fuel Cells CBS, National Energy Outlook 2016\*

McKinsey & Company

# Basic properties and Initial Safety Concerns

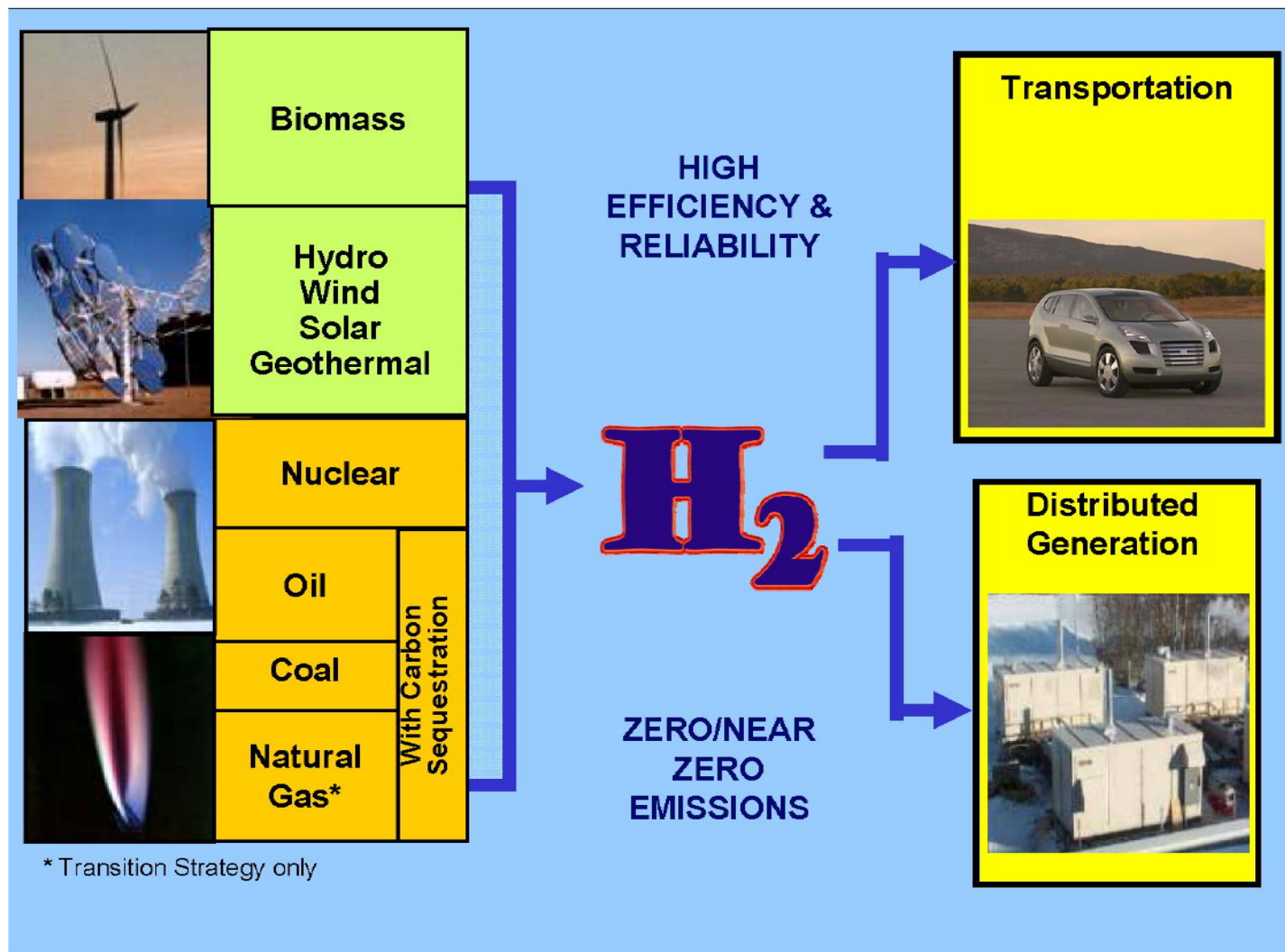
## Hydrogen (H<sub>2</sub>) is :

- the simplest, lightest and most abundant element in the universe.
- the first element in the periodic table and is a fundamental building block for many of the chemicals, materials and processes we use every day.
- a very small, diffusive molecule that is 14 times lighter than air.
- a non-toxic element that has been safely used in manufacturing for more than 90 years.
- is flammable (like all fuels) and safety systems at the H<sub>2</sub> stations (HRS) and in the H<sub>2</sub> vehicles (FCEV) are designed specifically for hydrogen's properties.
- an energy carrier as it can be used to store and transport energy, offering the possibility of greater diversification and sustainability in energy supply





# DIFFERENT ROUTES TO HYDROGEN



# Production Methods

Hydrogen production methods can be categorized based on their energy sources and environmental impact. So, here are the most common methods from :

## MATURE TECHNOLOGIES

### 1. Steam Methane Reforming (SMR) – Grey or Blue

**Process:** SMR is the most common method for producing hydrogen. It involves reacting natural gas (mostly methane) with steam at high temperatures to produce hydrogen and carbon monoxide. POX is a similar method but even less efficient than SMR.

### 2. Electrolysis – Green (if powered with RES)

**Process:** Electrolysis uses electricity to split water into hydrogen and oxygen.

### 3. Biomass Gasification – Brown

**Process:** Biomass (organic matter) is heated in the presence of a controlled amount of oxygen, producing a mixture of hydrogen, carbon monoxide, and  $\text{CO}_2$ .

### 4. Methane Pyrolysis - Turquoise (if powered with RES)

**Process:** Methane is split into hydrogen and solid carbon through high-temperature pyrolysis, without producing  $\text{CO}_2$ .

## EXPERIMENTAL PHASE

### 1. Photolysis (Photo-electrochemical Water Splitting) - Green Hydrogen

**Process:** Uses sunlight to split water into hydrogen and oxygen using special photo-sensitive materials.

Efficiency is lower compared to electrolysis (PEM, AEL, SOEL, AEM).

### 2. Thermochemical Water Splitting – Green or Pink

**Process:** Uses high-temperature heat, often from nuclear reactors or solar concentrators, to drive chemical reactions that split water into hydrogen and oxygen.

Efficiency expected to be high if coupled with high-efficiency heat sources.

### 3. Microbial Electrolysis Cells (MEC) - Green

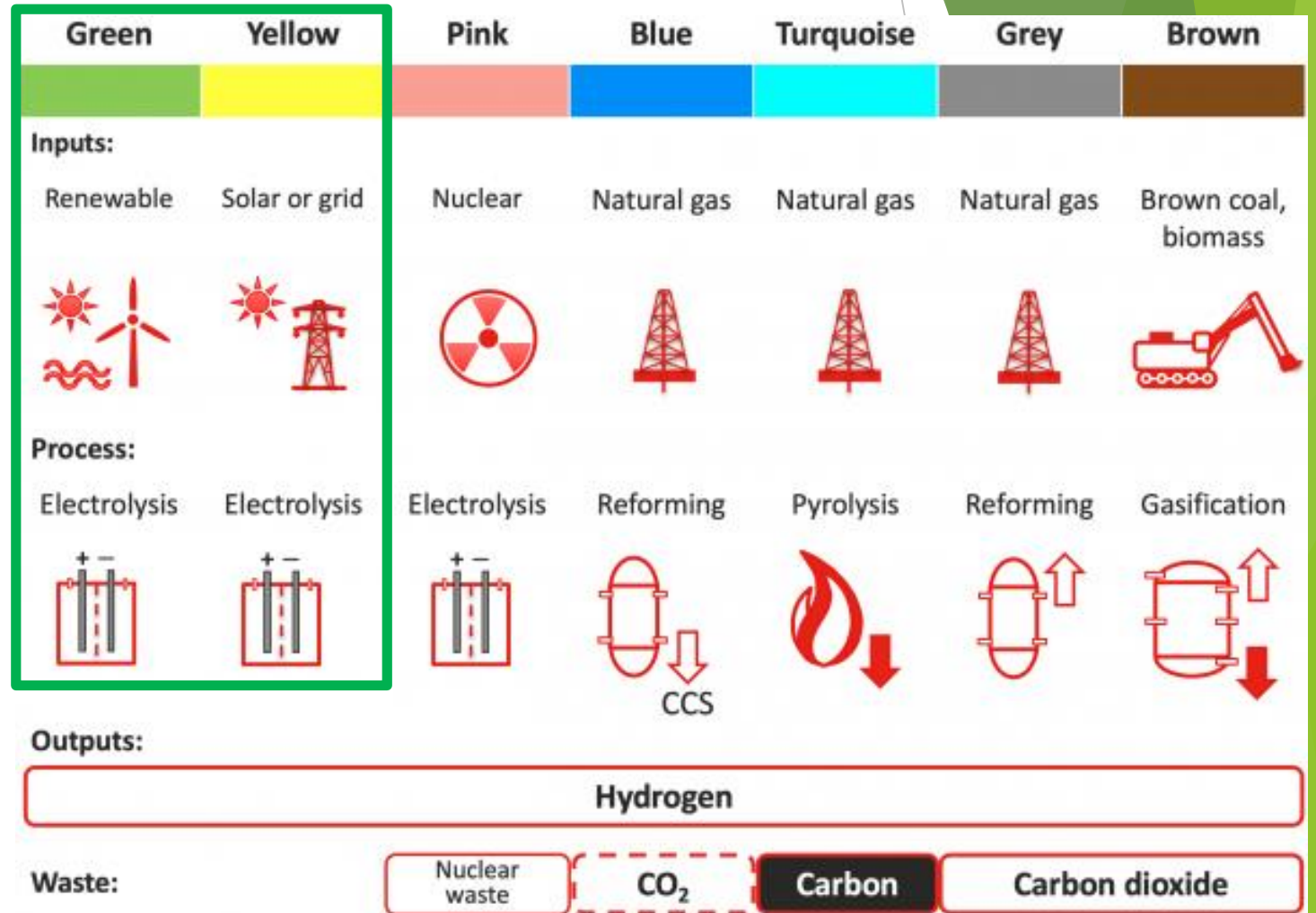
**Process:** Uses microbes and electricity to split water or wastewater into hydrogen.

Efficiency is low, but improving with research



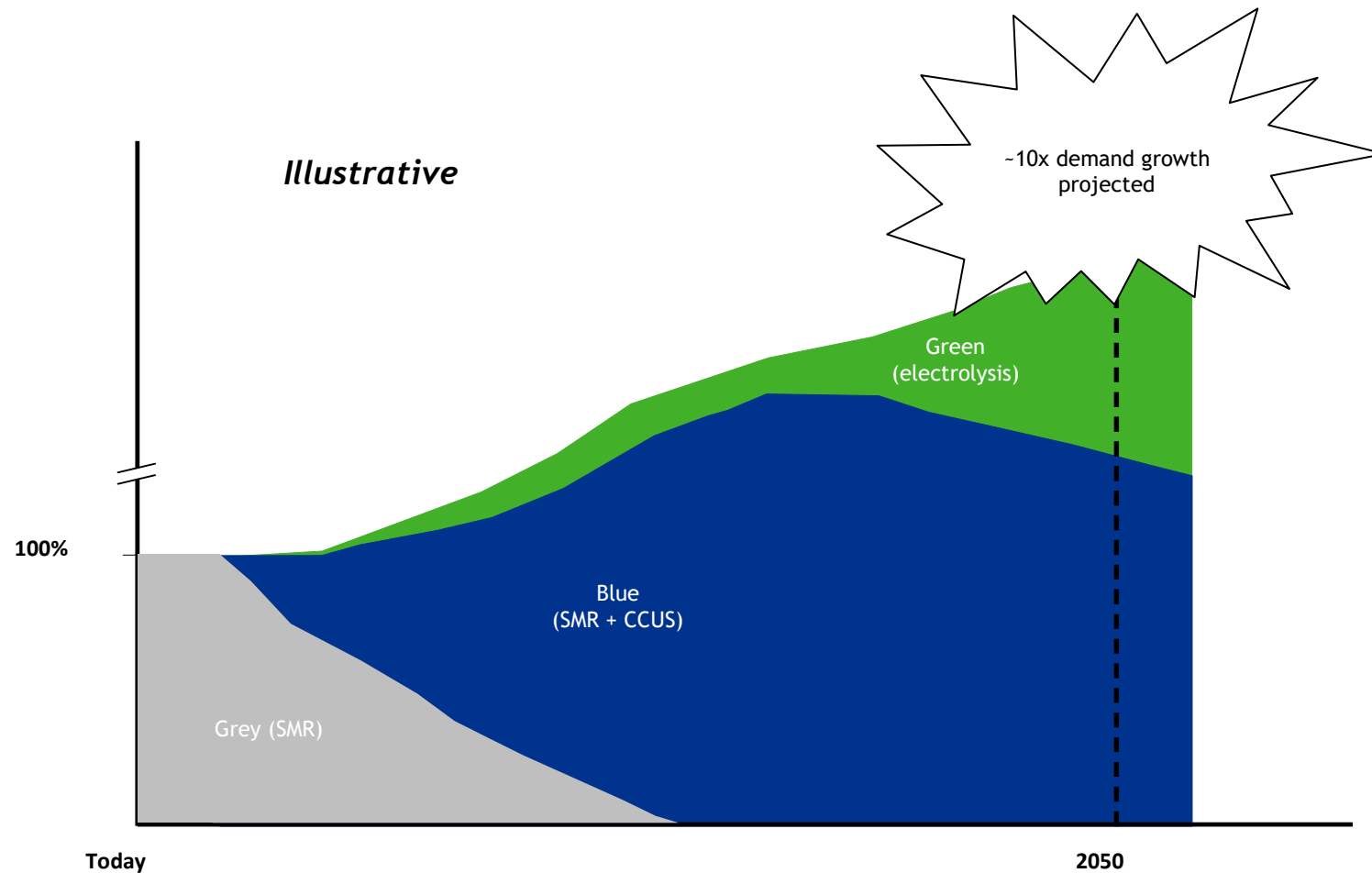
# Types (colors) of Hydrogen

- Each method offers different trade-offs in terms of :
  - ✓ Efficiency
  - ✓ Scalability
  - ✓ Environmental impact
- Green hydrogen**, though currently more expensive, is viewed as the future energy carrier, offering at the same time **zero-emission energy**.



# Hydrogen market capture strategies are customized by region

Hydrogen demand and mix over time



Source: Barclays, HSBC, Hydrogen Council

## Regional Hydrogen Strategy Drivers

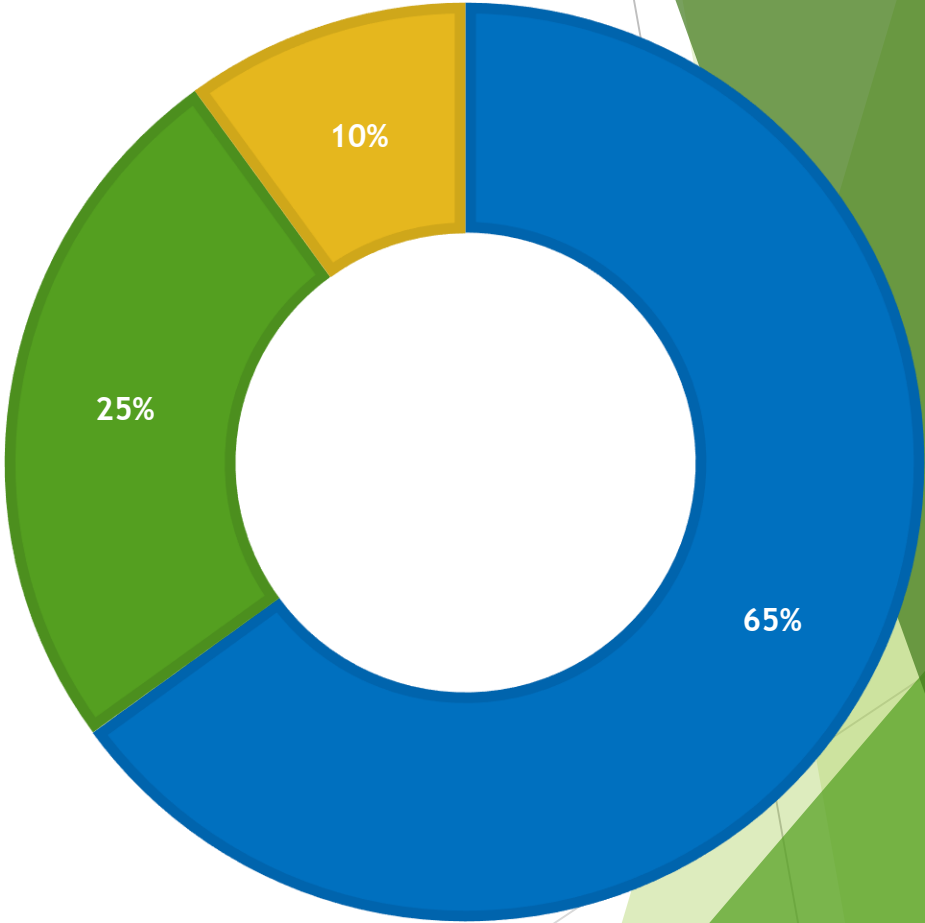
- Goals: 2050 net zero or similar
- Funding: Carbon fees or other
- Leverageable assets (blue)
  - H2 system
  - At-scale CCUS hub
- Leverageable assets (green)
  - Geologic storage
  - Low power prices

## Global Hydrogen Market Enablers

- Cost and supply chain improvements; e.g.,
  - Electrolyzers
  - Renewables
- H2 and renewable synergies

INDUSTRY SECTOR	KEY APPLICATIONS
CHEMICAL	<ul style="list-style-type: none"><li>• Ammonia</li><li>• Polymers</li><li>• Resins</li></ul>
REFINING	<ul style="list-style-type: none"><li>• Hydrocracking</li><li>• Hydrotreating</li></ul>
IRON & STEEL	<ul style="list-style-type: none"><li>• Annealing</li><li>• Blanketing gas</li><li>• Forming gas</li></ul>
GENERAL INDUSTRY	<ul style="list-style-type: none"><li>• Semiconductor</li><li>• Propellant fuel</li><li>• Glass production</li><li>• Hydrogenation of fats</li><li>• Cooling of generators</li></ul>

H2 DEMAND BY SECTOR



■ Chemical ■ Refining/Iron & Steel ■ General Industry

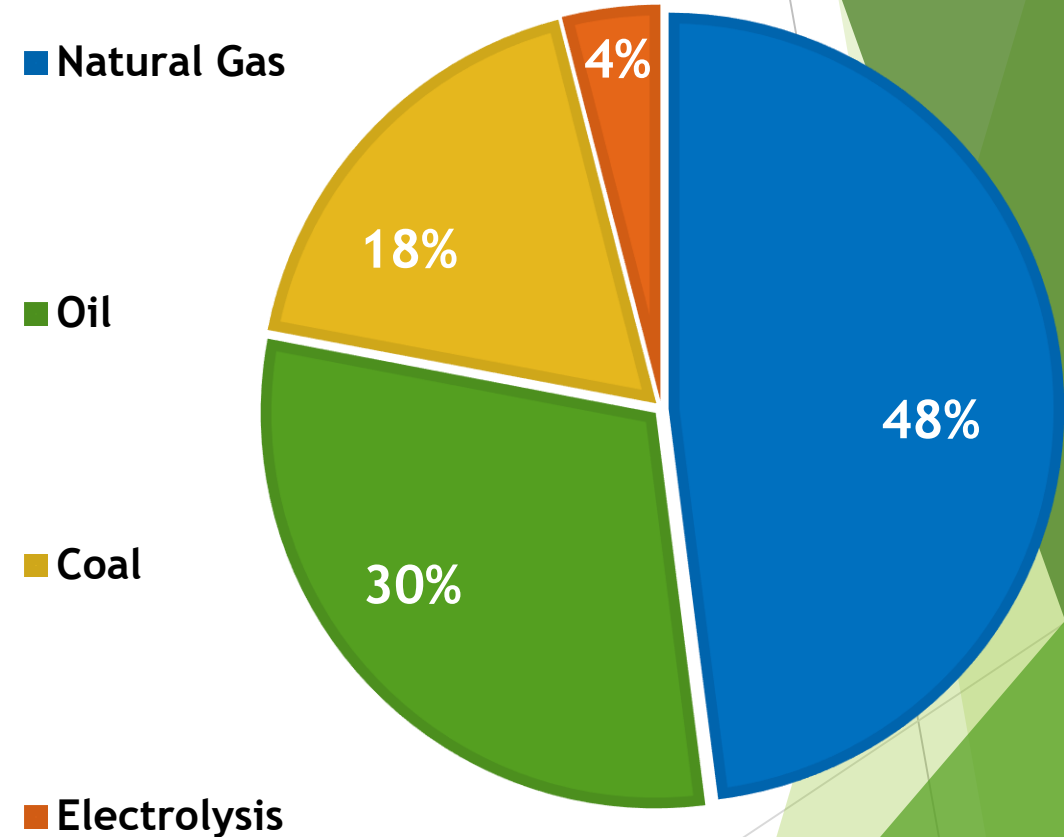
### Main routes for H2 production

- ▶ Hydrocarbon based production
- ▶ Water Electrolysis
- ▶ Biogas, CCS & novel routes

### Current situation

- 70 million metric tons of H2 produced in 2019, emitting in the process 830 million tons CO2
- About 95% of which is produced by fossil fuel based methods

### H2 PRODUCING SOURCE



# Steam Methane Reforming (SMR)

- ▶ SMR technology splits natural gas (NG) into mixture of hydrogen, CO and CO<sub>2</sub>
- ▶ Industries use SMR technology to deliver large quantities of H<sub>2</sub> and CO with remaining CO<sub>2</sub> being released to atmosphere
- ▶ The process is mature with systems capable of producing up to 100s of metric tonnes per day of H<sub>2</sub>/CO
- ▶ Currently over half the world's hydrogen is derived from NG feedstocks





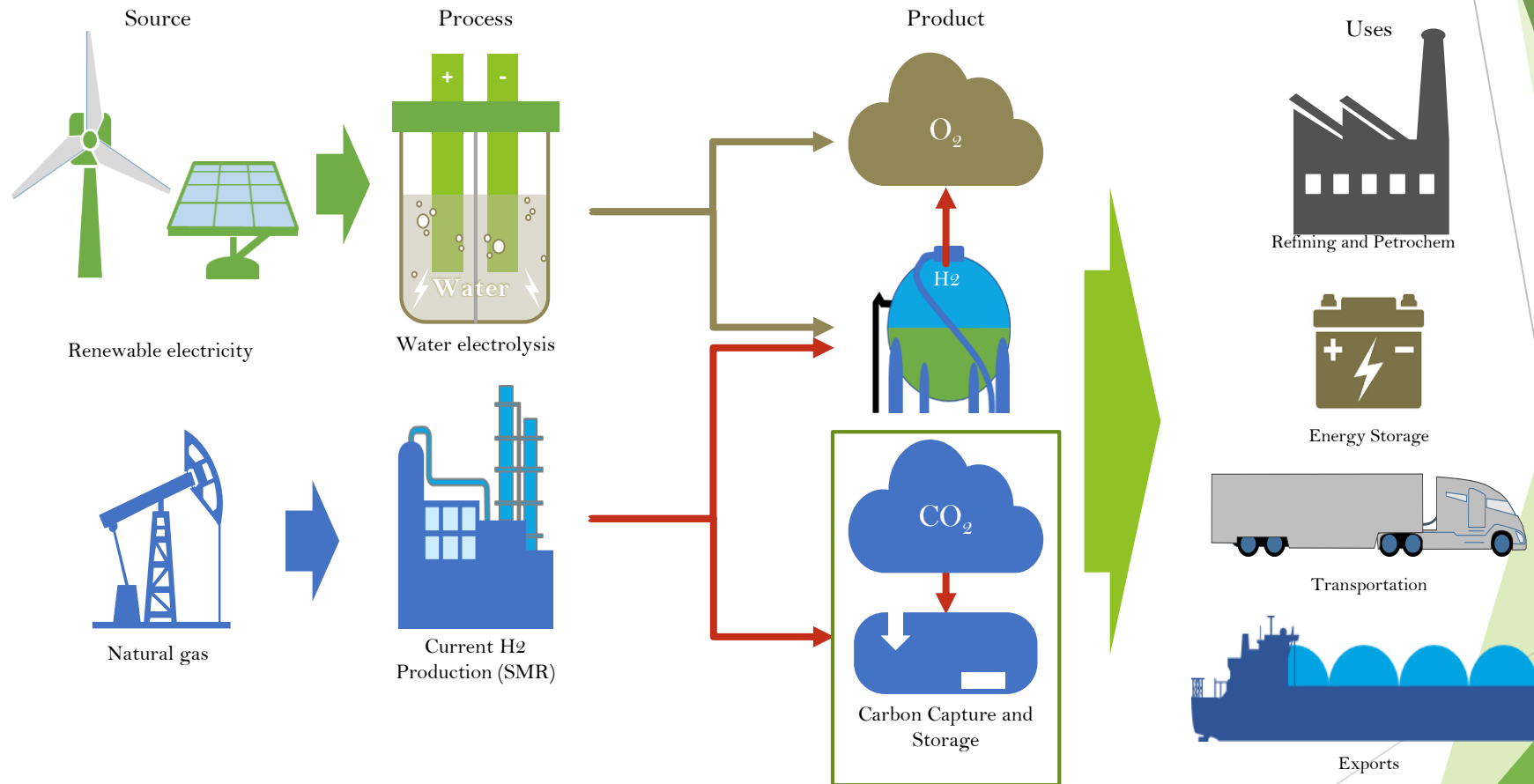
# SMR and Carbon Capture Utilisation and Storage (CCUS)

- ▶ Capturing the CO<sub>2</sub> produced by SMR can decarbonise the process
- ▶ Captured CO<sub>2</sub> can be stored in underground geological features or used
- ▶ **If CO<sub>2</sub> is captured/used, HYDROGEN IS BLUE**
- ▶ Many large scale CCUS projects are being developed internationally.



The **Acorn Project** will capture about 200,000 tonnes of CO<sub>2</sub> from the St Fergus Gas Terminal

# Typical Clean Hydrogen Production Options

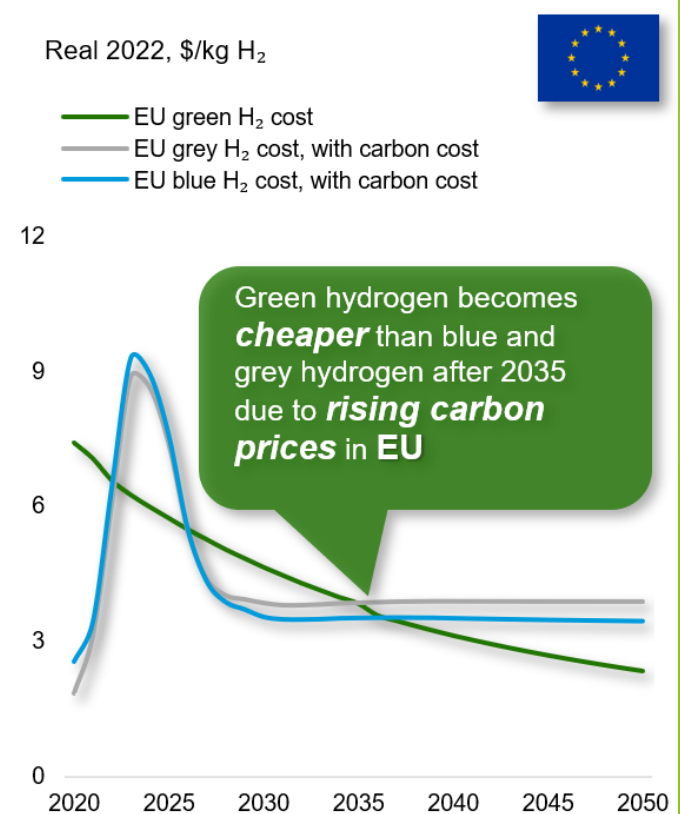
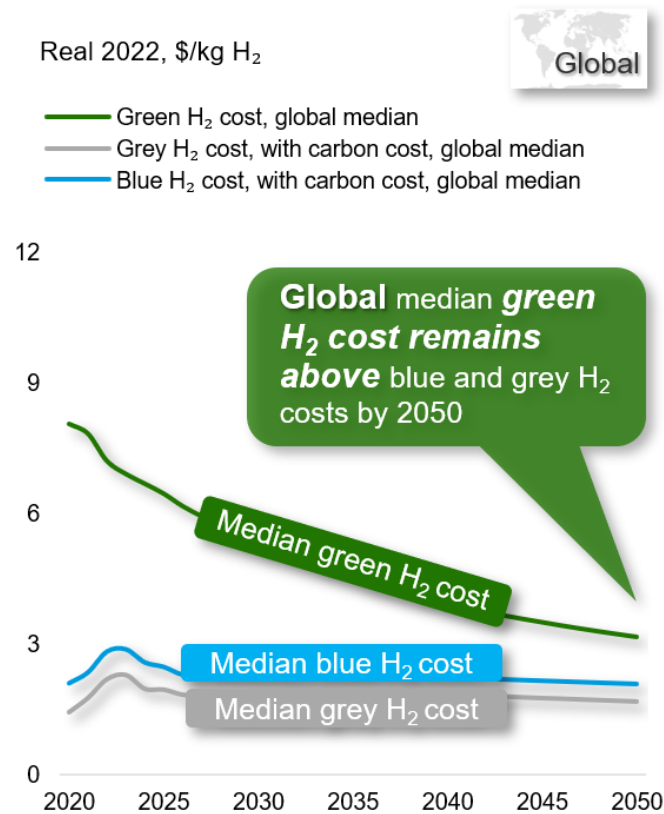


# What is the cost of Green H<sub>2</sub> ?

- ❖ The cost of hydrogen is heavily dependent on the way it is produced (Electrolysis or SMR).
- ❖ Blue/Grey hydrogen are a less costly option than Green Hydrogen but towards 2035 there is a break even point in EU
- ❖ **TODAY**, the cost of production for Green H<sub>2</sub> is approx. 5 – 7 € / kg H<sub>2</sub>
- ❖ **2035 TARGET PRICE**, is that Green H<sub>2</sub> reaches a price between 2-3 €/kgH<sub>2</sub>
- ❖ **TODAY it is sold** for 10 – 12 € / kg and looking for 2035 at a price of 5 - 6€ / kg H<sub>2</sub> and less reaching 2050.

## NOTE

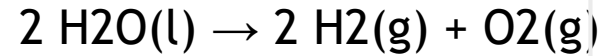
1 kg H<sub>2</sub> equals to approx. 4 lt of gasoline or petrol in terms of their calorific value.



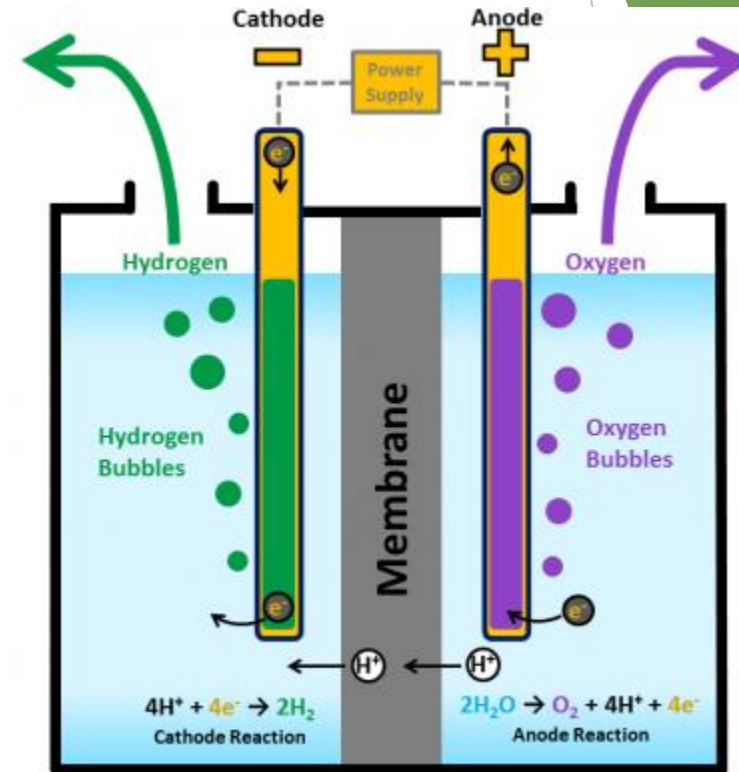
# Water ELECTROLYSIS

- One of the most promising methods for hydrogen production

## Reaction taking place:



- Requires a minimum energy of 39.4 kWh /kg of H<sub>2</sub>
- Has the potential to be carbon free, if powered by renewable sources

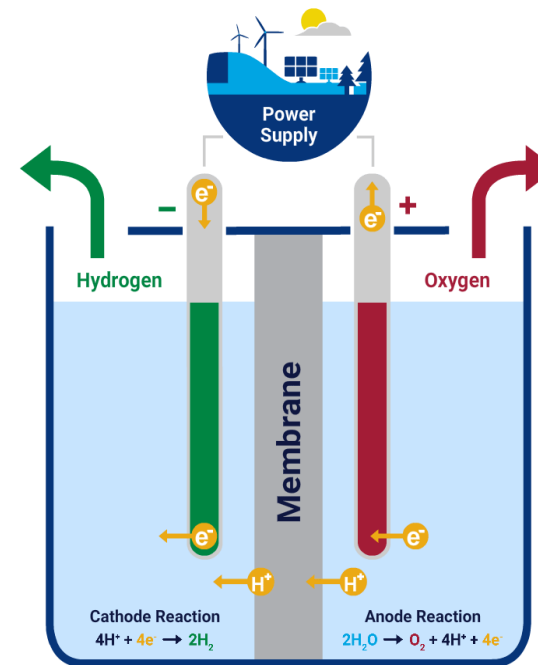


H<sub>2</sub> Production by water electrolysis

# Electrolysis - Production of Green Hydrogen

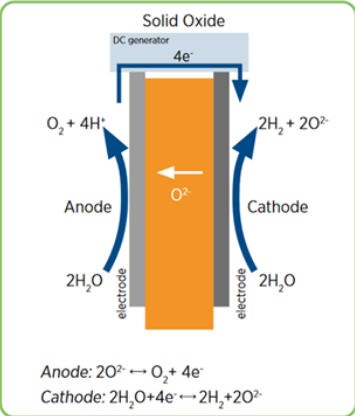
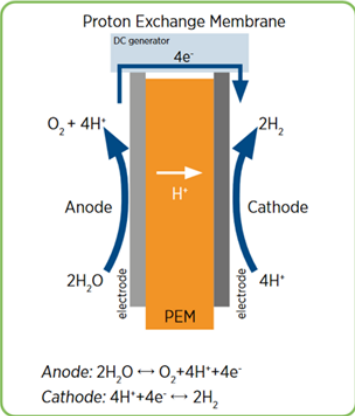
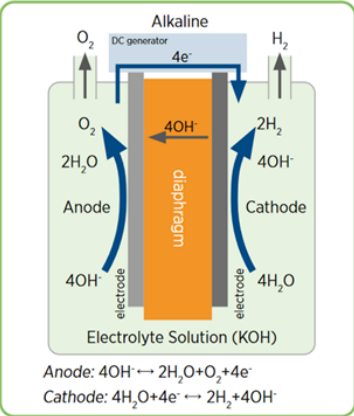
- First mentioned by Faraday, so not exactly modern technology.

- ▶ Green hydrogen produced.
- Electric current is passed through water to separate it in hydrogen and oxygen.
- Membranes to separate the gases.
- Alkaline
- PEM (and HT-PEM at about 180 C)
- Solid Oxide (High temperature > 600 C)





# Electrolyzer technologies



**Alkaline**

✓ Cheap

✗ Large equipment

**PEM**

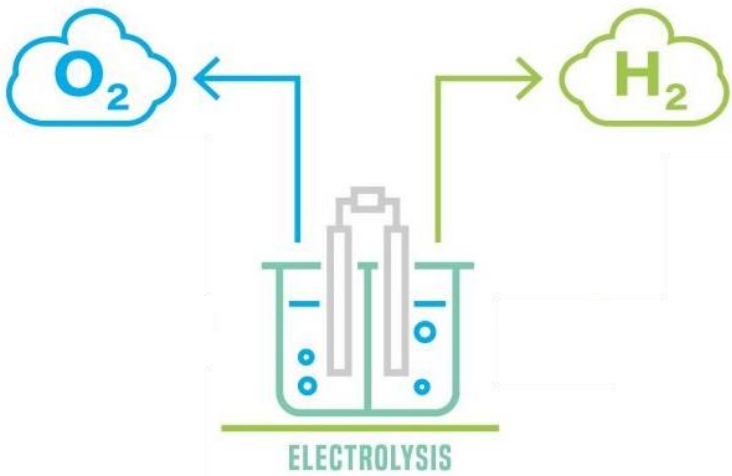
✓ Pure hydrogen

✗ Expensive catalysts

**Solid oxide**

✓ Reversible

✗ High temperature



	PEM	AEM	Alkaline	SOE
Electrolyte	PFSA membranes (e.g., Nafion)	Anion exchange ionomer	Aqueous potassium hydroxide	Yttria Stabilises Zirconia (YSZ)
Cathode	Platinum, Platinum - Palladium alloy	Nickel and Nickel alloys	Nickel, Nickel - Molybdenum alloy	Nickel/YSZ
Anode	Ruthenium oxide, Iridium oxide	Nickel, Ferrous, Cobalt oxides	Nickel, Nickel - Cobalt alloys	YSZ
Operating Temperature (°C)	50-80	50-60	60-80	500-850
Operating Pressure (Bar)	70	1-30	30	1-25
Stack Lifetime (h)	20-60k	-	60-100k	<10k
Technology Readiness	Commercialised	Large prototype	Matured	Demonstration
Cost	USD 1100-1800/kW		USD 500-1400/kW	USD 2800-5600/kW

	PEM-EL Polymer Electrolyte Membrane Electrolysis	AEM-EL Anion Exchange Membrane Electrolysis	A-EL Alkaline Electrolysis	HT-EL High-Temperature Electrolysis
Electrolyte	acidic	alkaline	alkaline	O <sup>2-</sup> - conducting solid (ceramic)
	solid (Polymer)		liquid	

# E- FUELS as a short term solution

- ▶ E fuels are synthesized from hydrogen and CO<sub>2</sub> from unavoidable emissions, producing e-methane , e-methanol or e- gasoline
- ▶ A short to mid-term plan is required, so we can make use of the existing infrastructure and a suitable solution could be energy rich e-fuels
- ▶ Gradual infiltration of hydrogen and green energy to carbon based fuels

Examples of E-fuel production:

$\text{CO}_2 + 4 \text{H}_2 \rightarrow \text{CH}_4 + 2 \text{H}_2\text{O}$  (e-Methane Production)

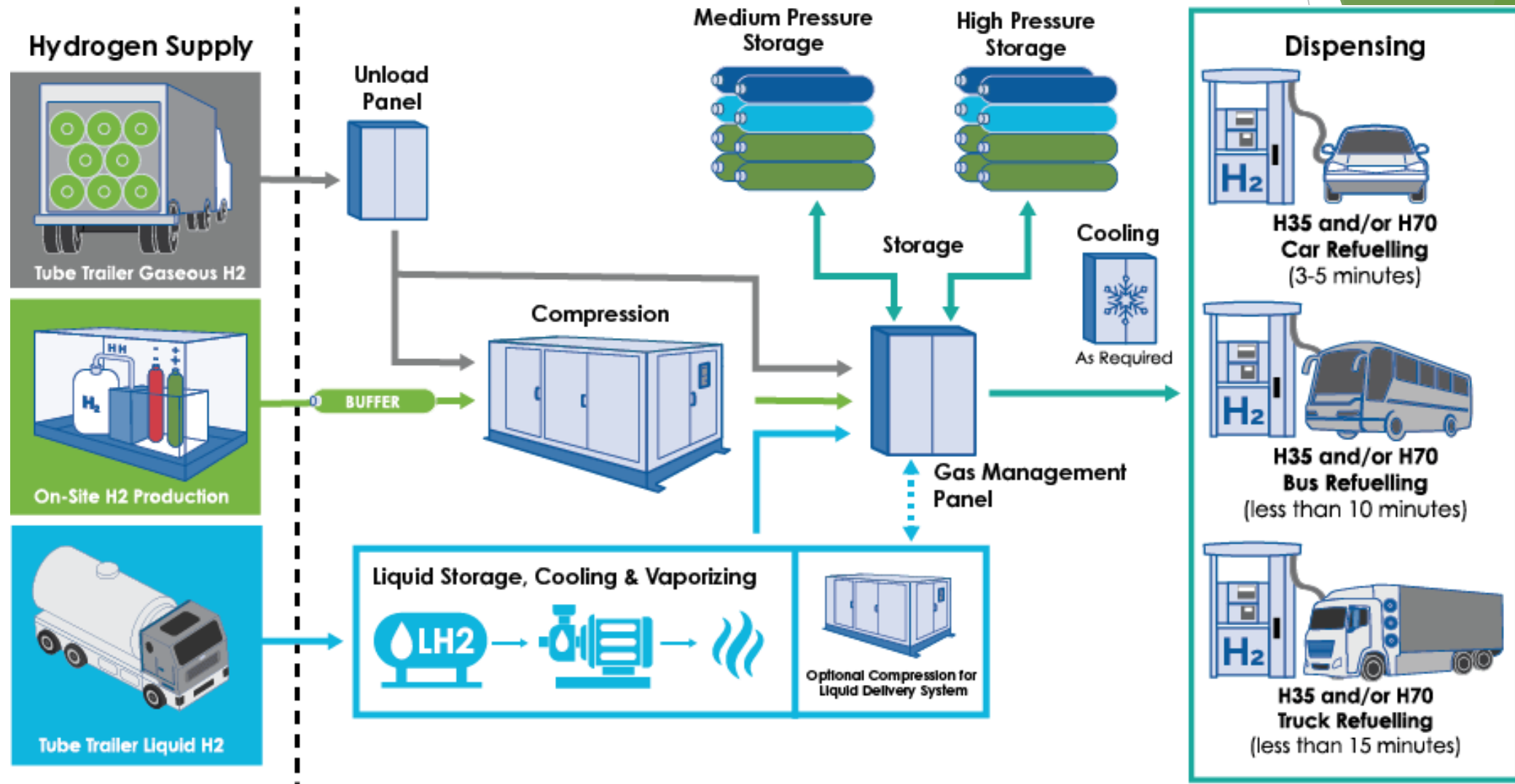
$\text{CO}_2 + 3 \text{H}_2 \rightarrow \text{CH}_3\text{OH} + \text{H}_2\text{O}$  (e-Methanol Production)

Fischer- Tropsch process for producing e- fuels:

$(2n + 1) \text{H}_2 + n \text{CO} \rightarrow \text{C}_n\text{H}_{2n+2} + n \text{H}_2\text{O}$

Overall balance of these processes is largely carbon neutral

# What is a HRS and how hydrogen is supplied?



Thanks for your interest!